Thalictricoside, a New Phenolic Compound from Thalictrum orientale

F. Zerrin Erdemgil^{a*}, Kemal Hüsnü Can Baser^a, Pinar Akbay^b, Otto Sticher^b, and Ihsan Çalis^c

- ^a Medicinal and Aromatic Plants and Drug Research Centre (TBAM), Anadolu University, 26470 Eskisehir, Turkey. Fax: 02223350750. E-mail: zerdemgi@anadolu.edu.tr
- b Department of Applied BioSciences, Swiss Federal Institute of Technology (ETH) Zurich, CH-8057 Zurich, Switzerland
- ^c Department of Pharmacognosy, Faculty of Pharmacy, Hacettepe University, TR-06100 Ankara, Turkey
- * Author for correspondence and reprint requests
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From the underground parts of *Thalictrum orientale* Boiss., a new phenolic compound **1** was isolated in addition to one known cyanoglycoside, lithospermoside (**2**). For the structure elucidation of all compounds, 1D- and 2D-NMR techniques (DEPT, COSY, HMBC, HSQC) and MS (HR-MALDI) were used. The structure of the new compound was established as 2-(4'-hydroxyphenyl)-nitroethane-4'-O-[β -xylopyranosyl-(1 \rightarrow 6)- β -glucopyranoside] (**1**).

Key words: Thalictrum orientale, Ranunculaceae Family, Thalictricoside, Lithospermoside

Introduction

The genus Thalictrum belongs to the Ranunculaceae family and is commonly known as 'meadow rue' (Schiff and Doskotch, 1970). The Thalictrum species are perennial herbaceous plants which are distributed in the temperate and tropical regions of the world (Bahadur and Shukla, 1983). Nine species of Thalictrum are known to grow in Turkey, namely; T. aquilegifolium L., T. flavum L., T. foetidum L., T. isopyroides C. A.Meyer, T. lucidum L., T. minus L., T. orientale Boiss., T. simplex L., T. sultanabadense Stapf, and three varieties of T. minus L. var. majus (Crantz) Crepin, var. minus L., and var. microphyllum Boiss. (Davis, 1965). Throughout the world, Thalictrum species have been used as stomachic, tonic, bitter, aperient, and for the treatment of snake bite, jaundice, rheumatism, etc. (Schiff and Doskotch, 1970). Thalictrum species have been extensively studied for their alkaloidal content. Several Thalictrum alkaloids have been reported to exhibit antitumor, antimicrobial, antitussive and hypotensive effects (Schiff and Doskotch, 1970). On the other hand, chemical studies on Thalictrum species have resulted in the isolation of cyanogenic glycosides, cycloartane glycosides, oleanane glycosides, flavone-C and O-glycosides (Sharples et al., 1972; Yoshimitsu et al., 1992; Yoshimitsu et al., 1994; Gromova et al., 1998; Wagner et al., 1971).

The first study in Turkey on the alkaloids of Thalictrum was achieved with T. lucidum in Turkey by Baytop and Berghmans (1975). Baser et al. have studied the alkaloids of several Thalictrum species (Baser, 1986; Baser and Kirimer, 1987; 1988; Baser and Ertan, 1990; Ertan and Baser 1997; Kirimer and Baser, 1991). Fifty eight alkaloids including twelve new compounds have been isolated and characterized by chromatographic and spectral studies (Erdemgil et al., 2001a). Recently, berberine, fuzitine and fangchinoline were isolated from Thalictrum orientale (Erdemgil et al., 2000; 2001b). Here we report on the isolation and characterization a new phenolic compound, thalictricoside (2) from T. orientale which is known as 'Kayaotu' in Nigde, Turkey. No pharmacological study and ethnomedical use of this plant has been reported (Erdemgil, 1999).

Materials and Methods

General experimental procedures

Optical rotation was measured on a Perkin-Elmer 141 polarimeter using a sodium lamp operating at 589 nm. UV spectra were performed on a Shimadzu UV-160A spectrophotometer. NMR measurements in pyridine-d₅ and CD₃OD were performed on Bruker AMX 300 and DRX 500 spectrometers operating at 300 and 500 MHz for ¹H and 75.5 MHz for ¹³C, respectively, using the XWIN NMR software package for the data acquisition and processing. HR-MALDI mass spectra were obtained on an IonSpec Ultima FTMS spectrometer (IonSpec, California, USA) by using 2,5-dihydroxybenzoic acid (DHB) as matrix.

Plant material

T. orientale was collected from Nigde, Ulukisla, Horoz village, in Turkey in June 1996, at an altitude of a *ca.* 1000 m. Voucher specimens have been deposited at the Herbarium of Faculty of Pharmacy, Anadolu University, Eskisehir, Turkey (ESSE 13296).

Extraction and isolation

The air-dried, coarsely powdered roots and rhizomes (1 kg) of T. orientale were extracted with 80 % MeOH in H₂O (61×2) under reflux for 8 h at 50 °C and filtered. The filtrate was concentrated to dryness in vacuo (260 g, yield 26 %). The extract was dissolved in H₂O (500 ml) and partitioned with CHCl₃ (150 ml \times 4) and BuOH (750 ml \times 3), respectively. The chloroform phase was evaporated to dryness in vacuo (1.59 g) and the butanol phase was concentrated by rotary evaporation yielding 17.0 g of butanolic extract. The butanol soluble part of the methanolic extract (17.0 g) was subjected to VLC using silica gel (Kieselgel 60, 30 g), employing CHCl₃, CHCl₃-MeOH (90:10) and CHCl₃-MeOH-H₂O (90:10:1→50:50:5) and MeOH as the eluents. 12 main fractions were collected (fraction A-L); A (26 mg), B (1.233 g), C (28 mg), D (148 mg), E (155 mg), F (146 mg), G (771 mg), H (948 mg), I (416 mg), J (1.554 g), K (4.142 g) and L (995 mg). Fraction I (416 mg) was applied to MPLC (column dimensions: $45 \times$ 2.5 cm) using Lichroprep RP-18 as a stationary phase and eluted with MeOH-H₂O mixtures (10-100 %) to give eighteen fractions (fraction volume; 100 ml). Fraction G (82 mg) was subjected to a Si gel (30 g) column using (EtOAc-MeOH-H₂O) (100:10:5; 200 ml; 100:15:5, 100 ml; 100:15:10, 100 ml; 100:20:10, 100 ml; 100:30:20, 100 ml) to yield compound 1 (frs. 19–20, 8.3 mg). Fraction K (4.142 g) was subjected to MPLC (column dimensions: 460 × 26 mm) using Lichroprep RP-18 as a stationary phase and eluted with MeOH-AcN-H₂0

mixtures with increasing polarity (20:5:75 \rightarrow 80:10:10) and MeOH to yield twenty four fractions (fraction volume; 250 ml). Fraction B and C were combined (2.332 g) and was subjected to normal phase silica gel (100 g) column chromatography using CHCl₃-MeOH-H₂O-NH₄OH mixture (61:32:5:1, v/v/v/v) yielding compound **2** (frs. 1–6; 57.79 mg).

Thalictricoside (1): Amorphous colorless powder; $[\alpha]_D^{20}$ – 57.9° (c = 0.1, MeOH); HR-MALDIMS m/z: 484 [M+Na]⁺, 413 [M-NO₂]⁺; UV $\lambda_{\rm max}$ (MeOH, nm): 224 (2.51), 272 (2.00); ¹H-NMR (500 MHz, CD₃OD) and ¹³C-NMR (CD₃OD, 75.5 MHz): Table I.

Lithospermoside (2): Amorphous colorless powder; $[\alpha]_D^{20} - 41^\circ$ (c = 0.1, MeOH); HR-MALDIMS m/z: 313.1041 [M+Na]+; UV $\lambda_{\rm max}$ (MeOH, nm): 207 (sh), 229, 275; IR $\nu_{\rm max}$ (KBr, cm⁻¹) 3346 (OH), 1722 (ester C=O), 1457 (aromatic ring); ¹H-NMR (300 MHz, CD₃OD) and ¹³C-NMR (CD₃OD, 75.5 MHz): Table II.

Results and Discussion

The butanol-soluble part of the methanolic extract of the underground parts of T. orientale was separately fractionated by vacuum liquid chromatography (VLC). After repeated chromatography (medium-pressure liquid chromatography = MPLC) of these fractions, a new compound (1) and the known compound lithospermoside (2) were isolated.

Fig. 1. N-containing glycosides (1, 2) from *Thalictrum orientale*.

The positive-ion HR-MALDI of compound 1 exhibited a pseudomolecular ion peak for [M + Na] $^+$ at m/z 484 which is compatible with the molecular formulae C₁₉H₂₇O₁₂N. The UV spectrum of 1 (λ_{max} 224 and 272 nm) was characteristic for phenolic compounds. Analysis of the ¹H-NMR spectrum of 1 (Table I) revealed the presence of a diglycosidic structure with a phenylethyl moiety. The anomeric proton resonances at $\delta_{\rm H}$ 4.32 (d, J=7.5 Hz) and 4.87 (d, J = 7.7 Hz) suggested the presence of two sugar units in 1. The complete assignments of all proton and carbon resonances were based on the DQF-COSY, HSQC and HMBC experiments. The DEPT and ¹³C-NMR spectra showed a total of 19 carbons consisting of four methylene, thirteen methine and two quaternary carbon resonances. Of the carbon signals eleven were attributed to the sugar moieties. This observation led us to consider the presence of hexose and pentose units. The signals belonging to the sugar units (Table I) were assigned by the help of COSY and HSQC experiments, indicating the presence of β -glucose and β -xylose units. In the

HMBC experiment, the long-range correlation between the C-6" ($\delta_{\rm C}$ 69.7) of glucose unit and the anomeric proton of xylose (H-1"; $\delta_{\rm H}$ 4.32) indicated the presence of β -xylopyranosyl- $(1\rightarrow 6)$ - β glucopyranoside. The remaining carbon and proton signals suggested the presence of a p-substituted phenylethyl moiety. Two aromatic signals observed as an AA'BB' system at $\delta_{\rm H}$ 7.07 and 7.19 $(J_{AB}=8.7 \text{ Hz})$ confirmed this assumption. Additionally, an isolated spin system observed at $\delta_{\rm H}$ 3.24 and 4.67 (each t, J = 6.9 Hz) was attributed to an ethylene bridge-taking place in between a phenyl ring and a heteroatom. The connectivities between two molecular fragments were resolved by an HMBC experiment. The long-range correlations between C-4 ($\delta_{\rm C}$ 158) of phenyl moiety and the anomeric proton of glucose ($\delta_{\rm H}$ 4.87, H-1") unit revealed the site of glycosidation; while C-1' $(\delta_C 131.6)$ of the phenyl moiety showed the longrange correlations to the methylene protons $(\delta_{\rm H} 4.67 \text{ and } 3.24; \text{H}_2\text{-1} \text{ and H}_2\text{-2}, \text{ respectively}) \text{ of }$ the ethylenic fragment. Positive ion HR-MALDI mass spectrum of 1 exhibited [M + Na]+ peak at

Table I. ¹H and ¹³C NMR spectroscopic data and HMBC correlations for thalictricoside (1), (¹H NMR, 500 MHz; ¹³C NMR, 75.5 MHz).*

C/H Atom	δ_{C} (ppm)	DEPT-135	$\delta_{\rm H}$ (ppm), J [Hz]	HMBC (from C to H)
Aglycone				
1	77.5	CH_2	4.67 t (6.9)	H-2', H-6', H ₂ -2
2	33.6	CH_2^2	3.24 t (6.9)	H_{2} -1
1'	131.6	C	_	H-2', H-6', H ₂ -1 H ₂ -2
2' 3'	130.8	CH	7.19 d (8.7)	, , , , , ,
3′	118.0	CH	7.07 d (8.7)	
4'	158.1	C	_	H-1", H-2', H-3', H-5', H-6'
5′	118.0	CH	7.07 d (8.7)	
6'	130.8	CH	7.19 d (8.7)	
Glucose			` /	
1"	102.1	CH	4.87 d (7.7)	
2"	74.8	CH	3.47 ^a	
3"	77.6	CH	3.45 ^a	
4"	71.4	CH	3.37 t (9.5)	
5"	77.4	CH	3.36 ddd (9.5/6.4/1.8)	
6"	69.7	CH_2	4.10 dd (11.7/1.8)	
			3.76 dd (11.7/6.4)	
Xylose			` ,	
1‴	105.3	CH	4.32 d (7.5)	$H_2-6''_{a,b}, H-5'''_{a,b}$
2‴	75.0	CH	3.19 dd (7.5/9.0)	
3‴	77.5	CH	3.27 t (9.0)	
4‴	71.2	CH	3.47 m	
5‴	66.9	CH_2	3.82 dd (11.4/5.3) 3.10 dd (11.4/10.3)	

^{*} All proton and carbon assignment are based on 2D NMR (DQF-COSY, HSQC and HMBC).

a Signal patterns are unclear due to overlapping.

m/z 484.141. The fragment ion appearing at m/z 413.3 [M-NO₂]⁺ suggested the presence of a NO₂ group on the ethylenic side chain. This assumption was also confirmed by the chemical shift value of one of the methylene signals observed at $\delta_{\rm H}$ 4.67 (H₂-2). This chemical shift value assigned to the methylene signal next to the NO₂ group was in good agreement with those reported for aliphatic nitro derivatives (Pretsch *et al.*, 1981). Based on

these results, the structure of **1** was established as 2-(4'-hydroxyphenyl)-nitroethane- $4'-O-[\beta-\text{xylopy-ranosyl-}(1\rightarrow 6)-\beta-\text{glucopyranoside}]$. For this novel structure thalictricoside was proposed as trivial name.

The spectroscopic data (Table II) for compound **2** were found to be identical with those reported for lithospermoside, a cyanoglucoside isolated previously from *T. rugosum* (Wu *et al.*, 1979).

Table II. 1 H and 13 C NMR spectroscopic data and HMBC correlations for lithospermoside (2) (1 H NMR, 500 MHz; 13 C NMR, 75.5 MHz).*

H-2, H-3, H-5, H-6, H-7,
H-3, H-6, H-7
H-2
H-2
H-6
H-1', H-2, H-7
H-2, H-6
H-2, H-6, H-7
H-6
)

^{*} All proton and carbon assignment are based on 2D NMR (DQF-COSY, HSQC and HMBC).

^a Signal patterns are unclear due to overlapping.

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